## Chem 4A Scolars Worksheet 10

## MO Diagrams and Hybridization

Hao Wu
Office Hour: 5:30-6:30pm email: haowu@berkeley.edu
course website: http://chem4a.haowu.org/

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## Questions:

1. Rank s, p, sp, sp<sup>2</sup>,sp<sup>3</sup> orbitals by their energies from lowest to highest.

2. The nitrogen atom in  $NH_3$  is  $sp^3$  hybridized, unlike  $CH_4$  one of it's hybridized atomic orbitals has slightly lower energy. Draw the electronic configuration of hybridized atomic orbitals.

3. Draw the lew is structure for  $\mathrm{CO_3}^{2\text{-}}.$  4. Draw the MO diagram for  ${\rm CO_3}^{2-}$ .